

OUR LAND AND WATER

Toitü te Whenua, Toiora te Wai

Measuring Actual Denitrification to Understand Nitrogen Loads through Aquifers

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Goal

Develop and validate an improved method for quantifying the extent of denitrification that actually occurs in groundwater systems, based on measurement of 'excess N2'.

Purpose

Knowledge of where denitrification occurs will help land use managers plan where intensive land use can/can't be carried out –

leading to reduced environmental impacts on our land and waterways.

Nitrate

Nitrate is the most pervasive groundwater contaminant

ca. 40% of long-term groundwater monitoring sites show abovenatural concentrations

Denitrification is a natural process for nitrate remediation

 $\mathsf{NO3}\,\rightarrow\,\mathsf{NO2}\,\rightarrow\,\mathsf{NO}_{\,(g)}\,\rightarrow\,\mathsf{N}_2\mathsf{O}_{\,(g)}\,\rightarrow\,\mathsf{N}_{2\,(g)}$



Why do we care?

Surface water

- Increased nitrogen in surface waterways can promote the growth of periphyton and degrading aquatic habitat and recreational values
- Elevated levels of nitrate and ammoniacal nitrogen can be toxic to fish and other animals.

Groundwater

- High levels of nitrogen in groundwater sourced drinking water can affect human health.
 - High levels of nitrate can pose a risk to babies less than six months who are formula fed and the unborn foetus of pregnant women. Adults with specific rare metabolic disorders may also be at risk
 - Groundwater that is rich in nitrate has the potential to elevate nutrient levels in the surface water it drains into causing the problems listed above.



Groundwater resources of New Zealand (MBIE Core funding)



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Current Research:

1. Isotope indicators for source and fate of nitrate and ammonium, from soil to freshwater impacts

2. Nitrogen, carbon and biomarker dynamics in soil and sediment

Dual isotope identification of nitrate sources and sinks

- Currently we can use natural isotope ratios to identify the N source.
- These ratios can be distorted by denitrification.
- Quantifying denitrification will allow for more accurate determination of source.



Excess nitrogen from denitrification

 $NO3 \rightarrow NO2 \rightarrow NO_{(g)} \rightarrow N_2O_{(g)} \rightarrow N_{2(g)}$



Next step

quantify the extent of denitrification in field trials (2018)

- Waikato Region
- Approx. 30 sites



collaboration



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- Chemistry conditions/pathways for denitrification to occur
- Cross calibration on 15 sites in the Horizons Region and 15 sites in the Waikato Region

